

IRIDOID GLUCOSIDES IN *AVICENNIA OFFICINALIS*

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Key Word Index—*Avicennia officinalis*; Verbenaceae; iridoids; avicennioside; 7-cinnamoyl-8-epiloganic acid; geniposidic acid; 2'-cinnamoyl-mussaenosidic acid.

Abstract—The new iridoids avicennioside and 7-cinnamoyl-8-epiloganic acid, as well as the known compounds geniposidic acid and 2'-cinnamoyl-mussaenosidic acid, have been isolated from the leaves of *Avicennia officinalis*. In the plant the acids are accumulated as salts. The taxonomic significance of these findings is discussed.

INTRODUCTION

The genus *Avicennia* has been classified differently by several authors. There are different opinions concerning (i) the systematic rank of *Avicennia* as a monotypic family Avicenniaceae [1–6], as a subfamily Avicennioideae [7, 8], or as a tribe Avicenniae [9, 10]; (ii) the phylogenetic relationship of *Avicennia* to the Verbenaceae [11, 12], Santalales [13], Celastrales [14], Dipterocarpaceae and Ancistrocladaceae [1]; and (iii) the subdivision of the genus into sections and the definition of the species. Schauer [9] and Moldenke [1], who recognizes 11 different *Avicennia* species, includes *A. marina* in the section *Upata* and *A. officinalis* in the section *Donatia*. Other authors such as Lam [15] and Schimper [16], however, recognize only three *Avicennia* species and unite *A. marina* with *A. officinalis*. Since the distribution of iridoids can be a useful character at the generic level, we examined the occurrence of iridoids in some *Avicennia* species. Previously we described the iridoids isolated from *A. marina* (Forsk.) Vierh. [17]. We have examined now a second species, *A. officinalis* L.

RESULTS

Hydrophobic adsorption chromatography of an alcoholic extract of the leaves from *A. officinalis* followed by LPLC and HPLC separations on RP material afforded four iridoid glycosides.

Compound 1 is a C₉-iridoid glucoside: UV (203 nm, log ϵ = 3.8), IR (1650 cm⁻¹) and ¹H NMR data for H-3 (δ 6.32, *d*) and H-4 (5.11, *dd*, $J_{3,4}$ = 6.5 Hz, $J_{4,9}$ = 1.5 Hz) indicated the presence of a 4-unsubstituted enol ether system of iridoids. The ¹H NMR spectrum also showed typical signals for one β -glucopyranosyl moiety. The assignments of the sugar protons were confirmed by a two-dimensional (2D) NMR spectrum (COSY 45). The aglycone part of the one-dimensional (1D) spectrum showed an AB system for two protons adjacent to hydroxyl groups, which could be assigned to H-6 and H-7 by the COSY 45 spectrum and NOE difference spectra (see Table 1). The 2D NMR showed small (J < 1 Hz) but significant *W*-type long-range couplings between H₃-10 and H-7 as well as H-9, which were not resolved in the 1D spectra. These couplings allowed the assignment of the signal at δ 4.03 to H-7 and they indicated a *trans*-relationship between Me-10 and H-7 as well as H-9. The signal at 3.63 showed a positive NOE to H-4 and could therefore be assigned to H-6. The coupling constant $J_{6,7}$ = 10 Hz indicated a *trans*-diaxial arrangement of H-6 and H-7, which was confirmed by the lack of any NOE

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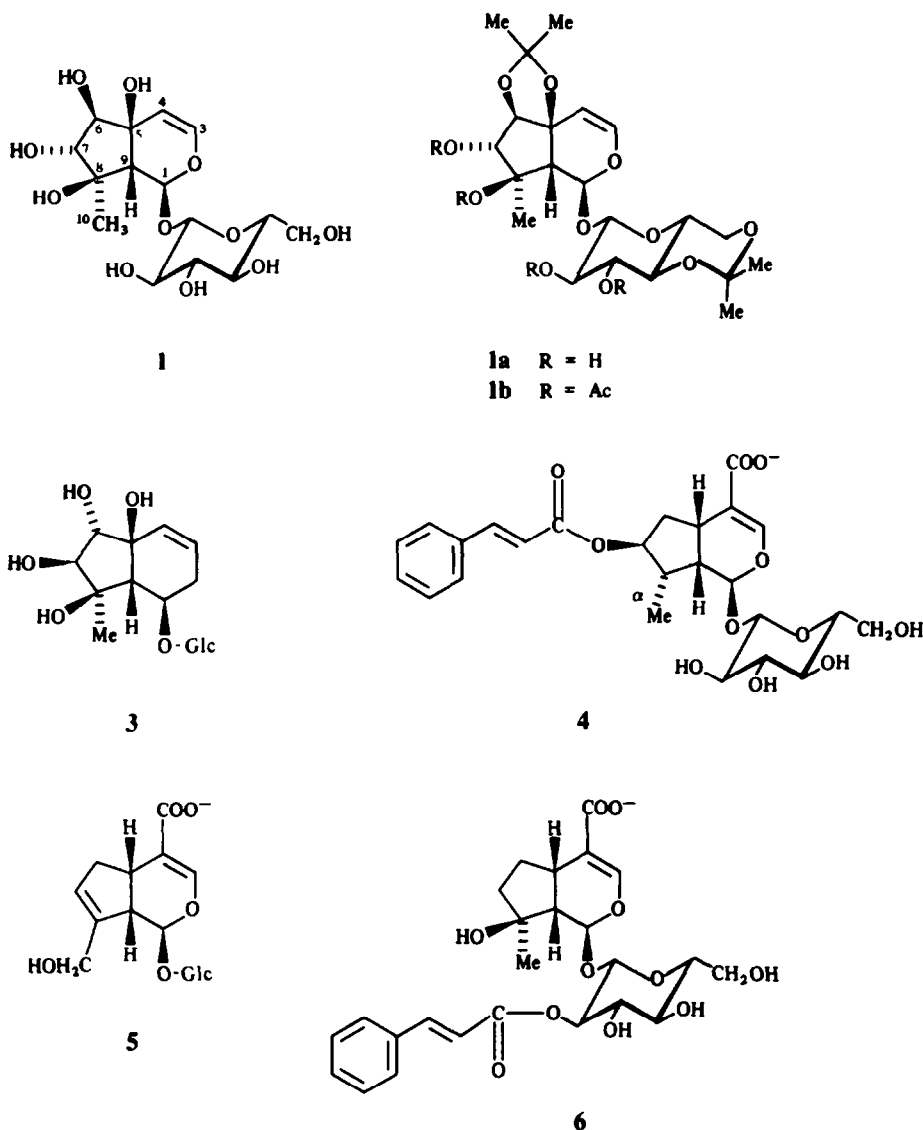
Table 1. Significant ¹H NMR spectral data (250.1 MHz/TMS) and NOEs of compounds 1, 1a and 1b

	1*			1a*			1b†	
	Irrad.	NOE‡		Irrad.	NOE‡		Irrad.	NOE‡
H-1	5.64	+	H-9, H-10, H-1'	5.47	—	5.81	+	H-9
H-3	6.32	+	H-4, H-10	6.35	—	6.32	—	
H-4	5.11	+	H-3, H-6	5.12	+	5.04	+	H-3, H-6
H-6	3.63	+	H-4, H-10	4.16	+	4.17	+	H-4
H-7	4.03	+	H-9	4.08	+	4.90	—	
H-9	2.41	+	H-7, H-1	2.47	+	3.26	+	H-1, H-7
H-10	1.15	+	H-1, H-3, H-6	1.10	+	—	—	
H-1'	4.64	—		4.68	—	4.84	—	

*In CD₃CN + 5% D₂O.

†In CDCl₃.

‡Positive NOE on the H-atoms listed.



between them. These arguments together with the observations that there were no signals for H-5 and H-8 and that the ^{13}C NMR shifts for C-8 and C-9 (Table 2) were very similar to other 8β -hydroxy- 8α -methyl-iridoids [18–20] led to structure 1 for this compound.

Structure 1 has already been claimed for 7α -hydroxyharpagide (2), a compound prepared semisynthetically from anthrithrhinoside [21, 22]. However, the published ^{13}C NMR data of that compound [19], especially the chemical shifts for C-6 and C-7, differed markedly from the data of our compound, which are substantiated by APT and selective decoupling experiments. Since we were unable to obtain an authentic sample of compound 2, we looked for further evidence to prove structure 1 and to exclude structure 3 for our compound. The ^1H NMR spectrum of the diacetone 1a showed a significant downfield shift of 0.53 ppm for H-6 whereas H-7 remained unchanged. Acetylation of 1a gave a tetraacetate (1b). In the ^1H NMR spectrum of this compound, H-1, H-9, and H-7 were shifted downfield by 0.33, 0.79 and

0.81 ppm, respectively, whereas the chemical shifts of H-4 and H-6 remained unchanged. These results are compatible only with one isopropylidene group being linked to O-5 and O-6 and, therefore, with a *cis*-relationship of OH-5 and OH-6 in 1. Moreover, the acetylation shifts for H-1 and H-9 corresponded well with the shifts observed for other 8β -hydroxy- 8α -methyliridoids [23], again confirming the configuration of 1 at C-8.

NOE difference spectra of 1, 1a and 1b (Table 1) showed positive NOEs between H-6 and H-4, H-6 and H₃-10, H-7 and H-9, thus proving that H-6 and Me-10 are on the concave α -side and H-7 is on the convex β -side of the iridoid moiety. Thus we have no doubt that 1 is the correct structure of our compound, for which we suggest the name *aviccennioside* in order to avoid confusion with compound 2. Although the base-catalysed ring opening of 7,8-epoxyiridoid glucosides has been studied extensively [22], and although a wide range of polyhydroxylated iridoid glucosides with structures similar to 1 are known, it seems to be difficult to determine the configuration of 2

Table 2. ^{13}C NMR spectral data of compounds 1, 2, 4 and 8-epiloganin

Carbon No.	1* (D ₂ O)	2† [21] (D ₂ O)	4‡ (DMSO- <i>d</i> ₆)	8-Epiloganin [24]* (D ₂ O)
1	93.12	93.18	93.31	96.5
3	141.31	141.13	(—)§	152.2
4	110.20	110.65	(—)	114.0
5	66.63	65.34	31.32	29.4
6	81.33	83.81	38.02	39.6
7	72.92	80.09	81.14	79.0
8	75.98	76.08	41.28	43.5
9	58.44	57.53	40.86	41.8
10	19.26	17.86	13.81	14.0
11	—	—	(—)	170.7
1'	99.96	99.91	98.09	99.1
2'	74.55	74.39	73.22	73.5
3'	77.42	77.45	76.78	76.6
4'	71.75	71.78	70.19	70.5
5'	78.25	78.23	77.15	77.1
6'	62.80	62.80	61.29	61.6

trans-Cinnamoyl moiety of 2: 118.35 (C- α), 128.23 (C-3''/C-5''), 128.79 (C-2''/C-6''), 130.29 (C-4''), 133.90 (C-1''), 144.20 (C- β), 165.94 (C=O).

*Capillary *d*₄-TSPNa/D₂O as external standard.

†Adjusted to $\delta_{\text{C-6}} = 62.80$ ppm.

‡TMS as internal standard.

§(—), not observed.

||Interchangeable.

with certainty by prediction of the reaction mechanism of the epoxide opening and the comparison of ^{13}C NMR data only.

The IR spectrum of compound 4 showed typical absorptions of the enol ether system of iridoids at 1635 cm^{-1} , an ester function at 1700 cm^{-1} and a carboxylate group at 1540 and 1400 cm^{-1} . Analysis by atomic absorption spectroscopy indicated that the counterion of the carboxylate was 70% sodium. Since we could not detect significant amounts of other metal ions, we suppose that the compound contained about 30% of the free acid. In the ^1H NMR spectrum of 4, the signals for H-7 and H-1' were covered by the HDO signal of the solvent, which could be suppressed by an inversion recovery technique with optimized recovery delay. With this method we obtained a completely resolved spectrum of 4, with only H-3 being of reduced signal intensity, typical for protons with no potent neighbours for dipolar relaxation. Especially in DMSO-*d*₆ the width at half height of the H-3 signal was markedly broadened due to the slow exchange process carboxylate \rightarrow carbonic acid. For the same reason, we assume that the signals for C-3, C-4 and C-11 were not observable in the ^{13}C NMR spectrum of 4.

The ^1H NMR and ^{13}C NMR spectra (Table 2) were similar to those of 7-*O*-(5-phenyl-2,4-pentadienyl)-8-epiloganin [17], but showed typical signals of a *trans*-cinnamoyl moiety. Compared with 8-epiloganin [24], the signal for H-7 was shifted downfield by *ca* 1 ppm in the ^1H NMR spectrum of 4. The ^{13}C NMR signal for C-7 was shifted downfield by 2.14 ppm whereas the signals for C-6 and C-8 were shifted upfield by 1.58 and 2.22 ppm, confirming that the ester group is linked to C-7. Methylation of the acid form of 4 with diazomethane followed by methanolysis yielded 8-epiloganin. After permethylation of 4 with methyl iodide in dimethyl

sulphoxide we obtained penta-*O*-methyl-8-epiloganin, which was identified by GC/MS [25, 26]. Compound 4 is therefore 7-*O-trans*-cinnamoyl-8-epiloganin.

The mass spectra of the TMSi derivatives of compounds 5 and 6 were identical with those of geniposidic acid and 2'-cinnamoyl-mussaenosidic acid, which we have previously isolated from *A. officinalis* [17]. The IR spectra, however, showed the typical absorption for carboxylate groups at 1540 and 1400 cm^{-1} . The main counterion, as determined by atomic absorption spectroscopy, was sodium. Geniposidic acid-Na (5) was identified by spectral analysis. The data were identical with those of the authentic sample [27]. The ^1H NMR spectrum of 5 also showed only a weak signal for H-3. As with 4, this is probably due to the lack of nearby protons for dipolar relaxation; by applying large pulse angles (e.g. 90°) the signal intensity of H-3 can be significantly reduced. 2'-Cinnamoyl-mussaenosidic acid-Na (6) was methylated with diazomethane. The reaction product proved to be identical with 2'-cinnamoyl-mussaenoside.

DISCUSSION

Avicennia officinalis and *A. marina* accumulate iridoid glucosides of the geniposide and lamiide types. In addition, *A. officinalis* contains avicennioside, an iridoid of the harpagide type. All these iridoids, which are characterized by an α -orientated methyl group at C-8 or a 7,8-double bond and a hydroxymethyl group at C-8, are probably derived from the same biogenetic precursors. They are formed via 8-epideoxyloganic acid [28–30], whereas iridoids with a β -orientated methyl group at C-8 are biosynthesized via deoxyloganic acid [31–33]. Iridoids of the lamiide and harpagide types are known from the subfamilies Viticoideae [34–36] and Verbenoideae

[37, 38] of the Verbenaceae and from other families of the Lamiales [39]. Geniposidic acid has been isolated from *Aloysia triphylla* (Verbenaceae) [27] and from some species of the Rubiaceae (Gentianiflorae) [40–42]. Thus the iridoid types in *A. marina* and *A. officinalis* show no closer relationship to any of the subfamilies of the Verbenaceae than to other families of the Lamiales. Morphologically, *Avicennia* differs from the typical verbenaceous plants in its incomplete four-celled ovary with a free central placenta, the orthotropous ovules, fruits with endosperm, a viviparous embryo, wood anatomy and pollen morphology [5, 6, 8, 43–45]. Our results combined with these morphological characters are more consistent with a relatively independent position of *Avicennia* amongst the Lamiales as a family or a subfamily [1, 11, 12] than with other opinions.

The iridoids of *A. marina* and *A. officinalis* are similar but they seem to be well distinguished. In *A. marina* only iridoid acids have been found, some of which are esterified with cinnamic or 5-phenyl-2,4-pentadienoic acid. From *A. officinalis* we exclusively obtained the salts of iridoid acids and their cinnamic acid esters. The C₉-iridoid avicennioside has been found only in *A. officinalis*. These results favour the recognition of two different species, as suggested by Moldenke [1].

EXPERIMENTAL

Plant material. Air-dried leaves of *A. officinalis* were obtained from wild plants growing in Sri Lanka. A voucher specimen (005-002) has been deposited at the herbarium of the Institut für Pharmazeutische Biologie, Freiburg.

Analytical methods. CC: Amberlite XAD 7 and XAD 2 (Serva), H₂O–MeOH (7:3)/(1:9); LPLC: LiChroprep-RP 8, size C (Merck), MeOH (10–60%), flow rate 120 ml/hr; HPLC: μ -Bondapak C₁₈, 9.4 mm \times 250 mm, MeOH (5–45%), flow rate 3 ml/min; TLC: silica gel 60, CH₂Cl₂–MeOH–H₂O (70:30:3–90:10:1). Spray reagent for iridoids: vanillin (3%) and H₂SO₄ (1%) in 100 ml EtOH followed by heating at 110° for 5–10 min. GC/MS: geniposidic acid-Na and 2'-cinnamoyl-mussaenosidic acid-Na were trimethylsilylated with TMSi-S (Serva) for 2 hr at room temp. OV-101 (1.5%) column, 1.2 m \times 2 mm i.d., temp. of column 250°/260°, He 40 ml/min, EIMS 30 eV; hepta-O-methyl-8-epiloganin: OV-17 (3%) column, 1.2 m \times 2 mm i.d., temp. of column 250°, He 38 ml/min, EIMS 30 eV.

Isolation procedure. The plant material (1 kg) was extracted by refluxing for 30 min once with 96% EtOH and twice with 70% EtOH. The combined extracts were evaporated *in vacuo* and chromatographed on a Celite column. Elution with *n*-hexane–CH₂Cl₂ (1:1) afforded a lipophilic fraction, which was discarded. Subsequent elution with CH₂Cl₂–MeOH (1:1) yielded a hydrophilic fraction, which was separated by CC on XAD-7 into three iridoid-containing fractions. The first fraction was rechromatographed on silica gel with CH₂Cl₂–MeOH–H₂O (70:30:3). Purification of the separated iridoids by LPLC and HPLC yielded 100 mg avicennioside and 70 mg geniposidic acid-Na. Rechromatography of the other two fractions on XAD-2 followed by LPLC and HPLC yielded 50 mg 2'-cinnamoyl-mussaenosidic acid-Na and 100 mg 7-cinnamoyl-8-epiloganin acid-Na.

Avicennioside (1). ¹H NMR (250.1 MHz, CD₃CN/D₂O): δ 1.15 (3H, s, H-10), 2.41 (1H, br s, H-9), 3.24 (dd, $J_{2/3} = 9$, $J_{1/2} = 8$ Hz, H-2'), 3.22 (dd, $J_{4/5} = 9$, $J_{3/4} = 8$ Hz, H-4'), 3.39 (ddd, $J_{5/6a} = 2$, $J_{5/6b} = 5.5$ Hz, H-5'), 3.42 (dd, H-3'), 3.63 (1H, d, $J_{6/7} = 10$ Hz, H-6), 3.65 (1H, dd, $J_{6a/6b} = 13$ Hz, H-6'b), 3.9 (1H, dd, H-6'a), 4.03 (1H, d, H-7), 4.64 (1H, d, H-1'), 5.11 (1H, dd, $J_{3/4} = 6.5$, $J_{4/5}$

= 1.5 Hz, H-4), 5.64 (1H, br s, H-1), 6.32 (1H, d, H-3).

5,6'-4,6'-Bis-O-isopropylidene-avicennioside (1a). Compound 1 (36 mg) was treated with Me₂CO and H₃[P(Mo₃O₁₀)₄] as described in ref. [46]. RP-HPLC of the residue with 50% MeOH yielded 4 mg 1a. ¹H NMR (250.1 MHz, CD₃CN/D₂O): δ 6.35 (1H, d, $J_{3/4} = 6.8$ Hz, H-3), 5.47 (1H, s, H-1), 5.12 (1H, dd, $J_{4/5} = 1.5$ Hz, H-4), 4.68 (1H, d, $J_{1/2} = 8$ Hz, H-1'), 4.16 (1H, d, $J_{6/7} = 7.5$ Hz, H-6), 4.08 (1H, d, H-7), 3.85 (dd, $J_{6ax/6eq} = 10.5$, $J_{5/6eq} = 5.5$ Hz, H-6'eq), 3.73 (t, $J_{5/6ax} = 10.5$ Hz, H-6'ax), 3.55 (t, $J_{3/4} = 9.3$, $J_{4/5} = 9.3$ Hz, H-4'), 3.43 (t, $J_{2/3} = 9.3$ Hz, H-3'), 3.27 (ddd, H-5'), 3.20 (dd, H-2'), 2.47 (1H, br s, H-9), 1.43, 1.41, 1.33, 1.30 (12H, 4s, isopropyl), 1.1 (3H, s, H-10).

Tetraacetate of 1a (1b). Compound 1a (4 mg) treated with Ac₂O (0.8 ml) in dry pyridine (0.8 ml) for 48 hr at room temp. and worked up as usual yielded 0.4 mg crude 1b. ¹H NMR (250.1 MHz, D₂O): δ 7.71 (1H, d, $J_{\alpha/\beta} = 16$ Hz, H- β), 7.58 (2H, d, $J_{4/9} < 1.4$ Hz, H-1), 5.13 (dd, $J_{2/3} = J_{3/4} = 9$ Hz, H-3'), 5.04 (dd, H-4), 4.96 (dd, $J_{1/2} = 7.8$ Hz, H-2'), 4.89 (d, $J_{6/7} = 6.3$ Hz, H-7), 4.84 (d, H-1'), 4.17 (d, H-6), 4.00 (dd, $J_{5/6eq} = 5.5$, $J_{6ax/6eq} = 10.5$ Hz, H-6'eq), 3.81 (t, $J_{5/6ax} = 10.5$ Hz, H-6'ax), 3.75 (t, $J_{4/5} = 9$ Hz, H-4'), 3.38 (ddd, H-5'), 3.26 (br s, H-9).

7-Cinnamoyl-8-epiloganin acid-Na (4). [α]_D²⁵ – 71.93° (H₂O; c 0.75); UV λ_{max}^{MeOH} (log ϵ) nm: 278 (4.03), 236 (sh), 222 (3.93), 217 (3.98), 205 (3.91). IR $\nu_{KBr}^{cm^{-1}}$: 1700, 1635, 1535, 1400. ¹H NMR (250.1 MHz, D₂O): δ 7.71 (1H, d, $J_{\alpha/\beta} = 16$ Hz, H- β), 7.58 (2H, m, H-2'), 7.43 (3H, m, H-3', H-4', H-5'), 7.02 (1H, d, $J_{3/5} < 1.5$ Hz, H-3), 6.54 (1H, d, H- α), 5.52 (1H, d, $J_{1/9} = 2.3$ Hz, H-1), 4.91 (1H, m, H-7), 4.72 (d, $J_{1/2} = 8$ Hz, H-1'), 3.90 (1H, dd, $J_{6a/6b} = 13$, $J_{6b/5} = 1.5$ Hz, H-6'b), 3.69 (1H, dd, $J_{6a/5} = 5.3$ Hz, H-6'a), 3.46 (t, $J_{3/4} = J_{2/3} = 9$ Hz, H-3'), 3.44 (ddd, $J_{4/5} = 9$ Hz, H-5'), 3.36 (t, H-4'), 3.25 (dd, H-2'), 3.08 (1H, dddd, $J_{5/9} = J_{5/6a} = 9$, $J_{5/6b} = 5.3$ Hz, H-5), 2.76 (1H, ddd, $J_{8/9} = 9$ Hz, H-9), 2.47 (1H, m, $J_{7/8} = 3.8$ Hz, H-8), 2.24 (1H, ddd, $J_{6a/6b} = 15$, $J_{6a/7} < 3$ Hz, H-6 α), 1.93 (1H, ddd, $J_{6\beta/7} = 5$ Hz, H-6 β), 1.00 (3H, d, $J_{8/10} = 7.5$ Hz, H-10).

7-Cinnamoyl-8-epiloganin (4a). Obtained from 4 by methylation with CH₂N₂. EIMS (solid probe, source 270°) 30 eV, *m/z* (rel. int.): 91 (100), 103 (73.76), 127 (24.62), 131 (52.42), 139 (11.2), 145 (38.05), 149 (19.11), 150 (17.57), 178 (10.42), 179 (16.78), 182 (13.85), 193 (10.05), 210 (11.89), 211 (4.11), 228 (3.54), 341 (0.31), 358 (0.14).

Penta-O-methyl-8-epiloganin (4b). Obtained from 4 by methylation with MeI in DMSO, as described by Hakomori [25] and Franke [26]. GC/MS *m/z* (rel. int.): 458 (0.11), 429 (0.11), 428 (0.07), 396 (0.07), 382 (0.07), 351 (0.29), 313 (0.11), 285 (0.59), 281 (0.65), 256 (0.44), 240 (0.40), 225 (2.54), 224 (1.84), 219 (5.16), 193 (11.29), 187 (99.99), 161 (2.14), 155 (19.11), 145 (6.78), 139 (2.17), 133 (1.77), 127 (13.17), 115 (6.42), 111 (71.43), 101 (43.21).

Atomic absorption spectroscopy. The salts of geniposidic acid (5) and 2'-cinnamoyl-mussaenosidic acid (6) were measured by AAS. They contained 3.6% and 2.43% Na, corresponding to 70 and 57%, respectively, of the theoretical value.

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REFERENCES

1. Moldenke, H. N. (1960) *Phytologia* 7, 123, 179, 259.
2. Correll, H. B. and Correll, D. S. (1982) *Flora of the Bahama Archipelago*.
3. Villiers, J.-F. (1975) *Flore Cameroun* 19, 59.

4. Chapman, V. J. (1976) *Mangrove Vegetation*. J. Cramer, Lehre.
5. Erdtman, G. (1945) *Sven. Bot. Tidskr.* **39**, 279.
6. Mukherjee, J. and Chanda, S. (1973) *Geophytology* **3**, 85.
7. Briquet, J. (1897) in *Die Natürlichen Pflanzenfamilien* (Engler, A. and Prantl, K., eds.) Vol. IVa, p. 132. Engelman, Leipzig.
8. Melchior, H. (1964) in *Engler's Syllabus der Pflanzenfamilien*, p. 435. Borntraeger, Berlin.
9. Schauer, J. C. (1847) in *Prodromus Syst. Nat. Regni Veg.* (De Candolle, A., ed.) Vol. XI, p. 658. Masson, Paris.
10. Hutchinson, J. (1969) *Evolution and Phylogeny of Flowering Plants, Dicotyledons*, Academic Press, London.
11. Thorne, R. F. (1983) *Nord. J. Bot.* **3**, 85.
12. Takhtajan, A. (1973) *Evolution und Ausbreitung der Blütenpflanzen*, Fischer, Stuttgart.
13. Tieghem, van M. P. (1898) *J. Botany* **12**, 345.
14. Dahlgren, R. (1975) *Bot. Not.* **128**, 119.
15. Lam, H. J. (1919) *The Verbenaceae of Malayan Archipelago*. De Waal, Groningen.
16. Schimper, A. F. W. (1891) *Botanische Mitteilungen aus den Tropen—3, Die Indomalayische Strandflora*. Verlag Gustav Fischer, Jena.
17. König, G. and Rimpler, H. (1985) *Phytochemistry* **24**, 1245.
18. Chaudhuri, R. K. Affi-Yazar, F. Ü. Sticher, O. and Winkler, T. (1980) *Tetrahedron* **36**, 2317.
19. Bianco, A., Caciola, P., Guiso, M., Iavarone, C. and Trogolo, C. (1981) *Gazz. Chim. Ital.* **111**, 201.
20. Damtoft, S., Jensen, S. R. and Nielsen, B. J. (1981) *Phytochemistry* **20**, 2717.
21. Scarpati, M. L., Guiso, M. and Esposito, P. (1968) *Gazz. Chim. Ital.* **98**, 177.
22. Davini, E., Iavarone, C. and Trogolo, C. (1983) *J. Org. Chem.* **48**, 2854.
23. Bianco, A., Guiso, M., Iavarone, C. and Trogolo, C. (1975) *Gazz. Chim. Ital.* **105**, 185.
24. Bianco, A. and Passacantilli, P. (1981) *Phytochemistry* **20**, 1873.
25. Hakomori, S. (1964) *J. Biochem.* **55**, 205.
26. Franke, A. (1985) Thesis, Freiburg.
27. Sauerbier, H. and Rimpler, H. (1986) *Biochem. Syst. Ecol.* **14**, 307.
28. Uesato, S., Ueda, S., Kobayashi, K., Miyauchi, M. and Inouye, H. (1984) *Tetrahedron Letters* **25**, 573.
29. Damtoft, S. (1981) *J. Chem. Soc. Chem. Commun.* 228.
30. Berg, T., Damtoft, S., Jensen, S. R., Nielsen, B. J. and Rickelt, L. F. (1985) *Phytochemistry* **24**, 491.
31. Balsevich, J. and Kurz, W. G. W. (1983) *Planta Med.* **49**, 79.
32. Damtoft, S., Jensen, S. R. and Nielsen, B. J. (1983) *J. Chem. Soc. Perkin Trans. 1*, 1943.
33. Uesato, S., Matsuda, S. and Inouye, H. (1984) *Chem. Pharm. Bull.* **32**, 1671.
34. Seghal, C. K., Taneja, S. C., Dhar, K. L. and Atal, C. K. (1983) *Phytochemistry* **22**, 1036.
35. Jacke, G. and Rimpler, H. (1983) *Phytochemistry* **22**, 1729.
36. Seghal, C. K., Taneja, S. C., Dhar, K. L. and Atal, C. K. (1982) *Phytochemistry* **21**, 363.
37. Milz, S. and Rimpler, H. (1979) *Z. Naturforsch. Teil C* **34**, 319.
38. Damtoft, S., Jensen, S. R. and Nielsen, B. J. (1979) *Taxon* **28**, 525.
39. Dahlgren, R., Jensen, S. R. and Nielsen, B. J. (1981) in *Phytochemistry and Angiosperm Phylogeny* (Young, D. A. and Siegler, S., eds) p. 149. Praeger, New York.
40. Guarnaccia, R., Madyastha, K. M., Tegtmeyer, E. and Coscia, C. J. (1972) *Tetrahedron Letters* 5125.
41. Bailleul, F., Rabaron, A., Koch, M. and Delaveau, P. (1979) *Planta Med.* **37**, 316.
42. Takeda, Y., Nishimura, H. and Inouye, H. (1975) *Phytochemistry* **14**, 2647.
43. Junell, S. (1934) *Symb. Bot. Upsal.* **4**, 1.
44. Mukerjee, J. and Chanda, S. (1978) *Trans. Bose Res. Inst.* **41**, 39.
45. Mukherjee, J. (1978) *J. Palynol.* **14**, 109.
46. Mori, H., Shibata, K., Tsuneda, K. and Sawai, M. (1967) *Chem. Pharm. Bull.* **15**, 460.